

## Ternary Ionic Compounds Worksheet File Type

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View full document. Name: Shiv Patel 10/2/20 Nomenclature Worksheet Binary and Ternary Ionic Compounds Directions: Complete the following table. Cation Anion Type Formula Unit Name Mg 2+ Cl 1- Binary MgCl 2 Magnesium Chloride NH 4 1+ S 2- Ternary (NH 4 ) 2 S Ammonium sulfide Na 3+ SO 2- Ternary Na 2 SO 3 Sodium sulfite Al 3+ N 3- Binary Al 3 N 3 Aluminum Nitrate Ca 2+ N 3- Binary Ca 3 N 2 Calcium Nitride K 4 CrO 2- Ternary K 2 CrO 4 Potassium chromate Ba 2+ CO 3 2- Ternary Ba 2 (CO 3 ) 2 ...

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Ternary Ionic Compounds Worksheet File Type Binary Ionic Compounds - Compounds with monatomic ions in it, a metallic ion and a nonmetallic ion. This allows only two types of atoms in the formula. Ex: Rb2O Ternary Ionic Compounds - Compounds with at least one polyatomic ion in it. This allows three or more types of atoms in the formula.

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Ternary Ionic Compounds Worksheet Ternary Ionic Compounds Worksheet A - side 2 1. Copper (II) sulfate 2. Potassium chlorate 3. Ammonium hydroxide 4. Potassium carbonate 5. Sodium sulfate 6. Potassium acetate 7. Ammonium chloride 8. Magnesium hydroxide 9. Zinc nitrate 10. Magnesium hydroxide 11. Calcium carbonate 12. Potassium hydroxide 13. Calcium hydroxide 14.

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Read Online Ternary Ionic Compounds Worksheet File TypeTernary Ionic Compounds Worksheet File Type Binary Ionic Compounds - Compounds with monatomic ions in it, a metallic ion and a nonmetallic ion. This allows only two types of atoms in the formula. Ex: Rb2O Ternary Ionic Compounds - Compounds with at least one polyatomic ion in it.

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Ternary Ionic CompoundsWWWooorrrkkkssshhheeeeett AAA -- d ssiiddee 222. Ternary Ionic CompoundsWWWooorrrkkkssshhheeeeett AAA --d ssiiddee 222. Directions: Circle or underline the polyatomic ion in each compound then write the correct name or formula. Don't forget to use roman numerals on transition metals. 1.

[Ternary Ionic CompoundsWWWooorrrkkkssshhheeeeett AAA -- d ...](#)

Ternary Ionic Compounds Worksheet Ternary Ionic Compounds Worksheet A - side 2 1. Copper (II) sulfate 2. Potassium chlorate 3. Ammonium hydroxide 4. Potassium carbonate 5. Sodium sulfate 6. Potassium acetate 7. Ammonium chloride 8. Magnesium hydroxide 9. Zinc nitrate 10. Magnesium hydroxide 11. Calcium carbonate 12. Potassium hydroxide 13. Calcium hydroxide 14.

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Ternary Ionic Compounds Worksheet A - side 2 1. Copper (II) sulfate 2. Potassium chlorate 3. Ammonium hydroxide 4. Potassium carbonate 5. Sodium sulfate 6. Potassium acetate 7. Ammonium chloride 8. Magnesium hydroxide 9. Zinc nitrate 10. Magnesium hydroxide 11. Calcium carbonate 12. Potassium hydroxide 13. Calcium hydroxide 14. Potassium carbonate 15. Aluminum phosphate 16.

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10/27/2020 Bellringer. Click here for the video on Ternary Ionic Compounds.. Click here for Ternary Ionic Compounds Worksheet.. Just do page 1 of the worksheet. This is due by Friday, October 30th. This information you will find in the Nomenclature slides and notes.

[Ternary Ionic Compounds with Polyatomic Ions 10/27/2020](#)

Worksheet: More Binary and Name \_\_\_\_ Ternary Compounds CHEMISTRY: A Study of Matter © 2004, GPB 6.7a KEY Write the formula of the compound formed from the following pairs of ions: 1. Li+, O2- \_\_\_\_ 2. Ag+, OH- \_\_\_\_ 3. Mg2+, SO 4 2- \_\_\_\_ 4. NH 4 +, CO 3 2- \_\_\_\_ 5. Ca2+, P3- \_\_\_\_ 6. Sn4+, O2-