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CHAPTER 9 REVIEW Stoichiometry

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answer on the line to the left. Show all
your work in the space provided. 1.

88% The actual yield of a reaction is
22 g and the theoretical yield is 25 g.

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Calculate the percentage yield. 2. 6.0
mol of N_2 are mixed with 12.0 mol of
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ANSWER Answer the following
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molecule is caused by an uneven distribution of electrons between the oxygen and hydrogen atoms. 9. The concentration of H ions determines whether a solution is acidic or basic.

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